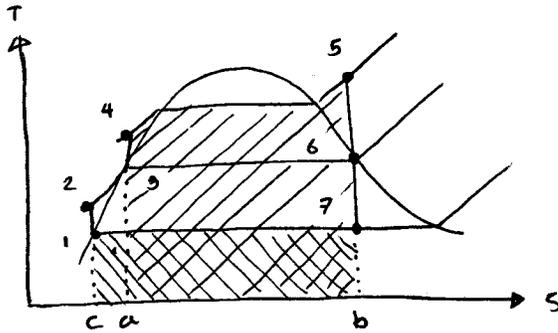


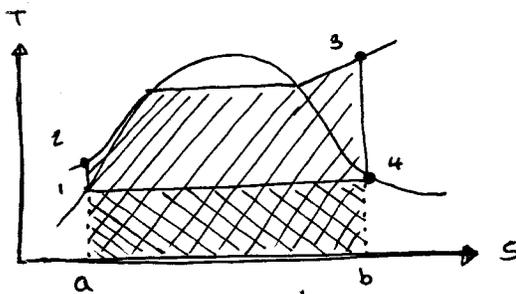
OCT. 23/18

Feedwater heaters :



$$A_{a45b} = Q_H$$

$$A_{c17b} = Q_L$$



$$Q_H = A_{a23b}$$

$$Q_L = A_{a14b}$$

Standard Rankine Cycle

Example:

(9.40)

FWH receives steam at 1 MPa, 200°C - From the turbine @ 1 MPa, 100°C water from the feed water line. Required fraction of extraction flow?

$$y = \dot{m}_6 / \dot{m}_5$$

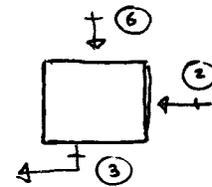
$$\dot{m}_7 = (1-y) \dot{m}_5$$

$$\dot{m}_2 h_2 + \dot{m}_6 h_6 = \dot{m}_3 h_3$$

$$(1-y) \dot{m}_5 h_2 + y \dot{m}_5 h_6 = \dot{m}_5 h_3$$

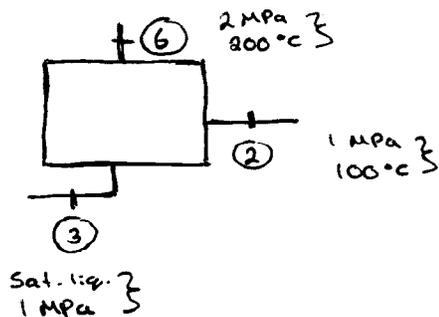
$$(1-y) h_2 + y h_6 = h_3$$

$$\Rightarrow y = \frac{h_3 - h_2}{h_6 - h_2}$$



$$\dot{m}_3 = \dot{m}_5$$

$$\dot{m}_2 = \dot{m}_7$$



$$y = ?$$

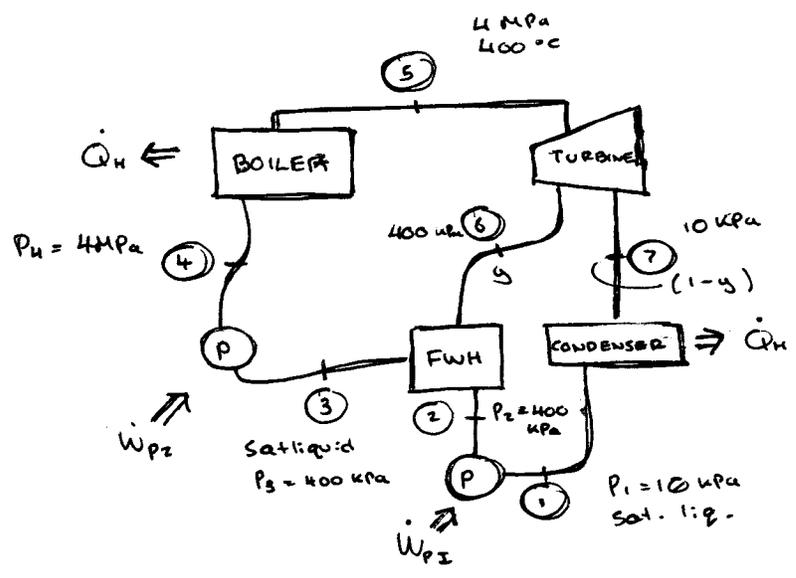
$$y = \frac{h_3 - h_2}{h_6 - h_2}$$

$$y = \frac{(762.79 - 419.02)}{(2827.86 - 419.02)} = 0.1427$$

$$\dot{m}_5 = 1 \text{ kg/s} \rightarrow \dot{m}_6 = 0.1427(1) = 0.1427 \text{ kg/s}$$

Example:

Steam leaves boiler and enters the turbine at 4MPa, 400°C. After expansion to 400 kPa, some of the steam is extracted from the turbine to open FWH. FWH pressure is 400 kPa water leaving is sat. liquid at 400 kPa, liquid from condenser: 10 kPa



$\eta_{th} = ?$

$$w_{P1} = v(P_2 - P_1) = 0.00101(400 - 10) = 0.4 \text{ kJ/kg}$$

$$w_{P1} = h_2 - h_1 \Rightarrow h_2 = h_1 + w_{P1} = 191.8 + 0.4 = 192.2 \text{ kJ/kg}$$

Turbine:

$$m_5 h_5 = m_6 h_6 + m_7 h_7 + \dot{W}_T$$

$$h_5 = y h_6 + (1-y) h_7 + w_t$$

$$w_t = h_5 - y h_6 - (1-y) h_7$$

$h_5 : P_5 = T_5 : h_5 = 3213.51$
 $h_6 : P_6 = S_6 = S_5 : h_6 = 2685.60$
 $h_7 : P_7 = S_7 = S_6 : h_7 = 2144.10$

$w_t = 980.06 \text{ kJ/kg}$

$$w_t = 3213.51 - y(2685.6) - (1-y)(2144.1)$$

FWH: $m_6 h_6 + m_2 h_2 = m_3 h_3 \Rightarrow y = \frac{h_3 - h_2}{h_6 - h_2}$

$$y = \frac{(604.7 - 192.2)}{(2685.6 - 192.2)} \Rightarrow y = 0.165$$

$$W_{net} = W_t - W_{pI} - (1-y)W_{pI}$$

$$W_{net} = 980.06 - (3.9) - (1-0.165)(0.4) = 975.8 \text{ kJ/kg}$$

$$W_{pII} = v_3(P_3 - P_4) \Rightarrow (0.001084)(400 - 4000) = -3.9 \text{ kJ/kg}$$

$$W_{pI} = h_3 - h_4 \Rightarrow h_4 = 608.6 \text{ kJ/kg}$$

Boiler:

$$q_H = h_5 - h_4 \Rightarrow 3213.51 - 608.6 = 2605 \text{ kJ/kg}$$

$$\eta_{th} = \frac{W_{net}}{q_H} = \frac{975.8}{2605} = 0.375 \text{ or } 37.5\%$$

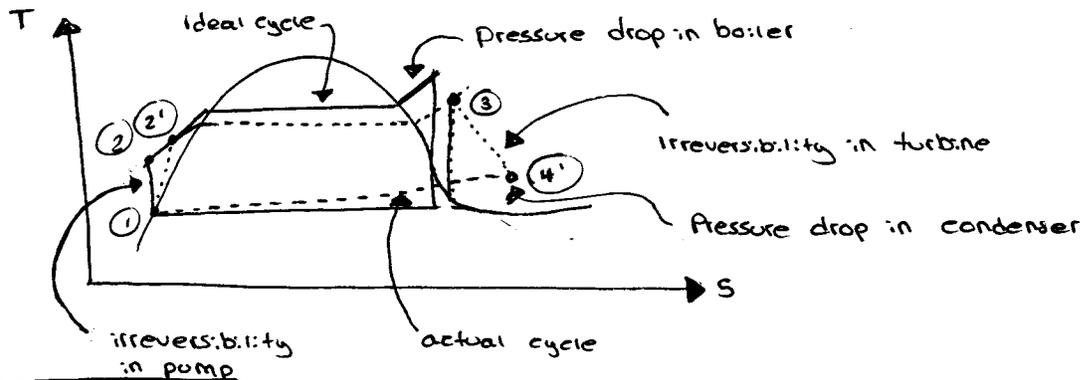
$$\eta_{th} = \frac{W_{net}}{q_H} = \frac{q_H - (1-y)q_L}{q_H}$$

END

Example - From picture

Oct. 25 / 18

Deviation of actual cycles From ideal cycles



Ex. (9.5)

Example:

A steam power plant operates on a cycle w/ pressures and temp. as designed in figure. Thermal efficiency is 86%, efficiency of pump is 80%.

Determine the thermal efficiency of this cycle.

$$\eta_{th} = \frac{W_{net}}{Q_H} \quad ; \quad W_{net} = W_{turbine (actual)} - W_{pump (actual)}$$

$$\eta_{turbine} = 0.86 \quad \rightarrow \quad \eta_{turbine} = \frac{W_{t,a} (actual)}{W_{h,s} (isentropic)} = \frac{h_5 - h_{6a}}{h_5 - h_{6s}}$$

$$\eta_{pump} = 0.80$$

$$W_{ta} = \eta_{turbine} (h_5 - h_{6s})$$

From table, $h_5 = 3169.1 \text{ kJ/kg}$

$$S_5 = S_{6s} = 6.7235 \text{ kJ/kg}$$

@ 10 kPa: $S_{6s} = S_{f/10kPa} + x_6 S_{fg/10kPa}$

$$x_6 = 0.8098$$

$$h_{6s} = h_{f/10kPa} + x_6 h_{fg/10kPa}$$

$$h_{6s} = 2129.5 \text{ kJ/kg}$$

$$W_{t,ac} = 0.86 (3169.1 - 2129.5)$$

$$W_{t,ac} = 894.1 \text{ kJ/kg}$$

$$\eta_{pump} = \frac{W_{ps}}{W_{pa}} = \frac{h_{2s} - h_1}{h_{2a} - h_1}$$

$$W_{pa} = \frac{h_{2s} - h_1}{\eta_{pump}} = \frac{v_1 (P_2 - P_1)}{\eta_{pump}}$$

$$\Rightarrow \frac{0.00101 (5000 - 10)}{(0.8)} = 6.3 \text{ kJ/kg}$$

$$W_{net,ac} = W_{t,ac} - W_{p,ac} \Rightarrow 894.1 - 6.3 \text{ kJ/kg}$$

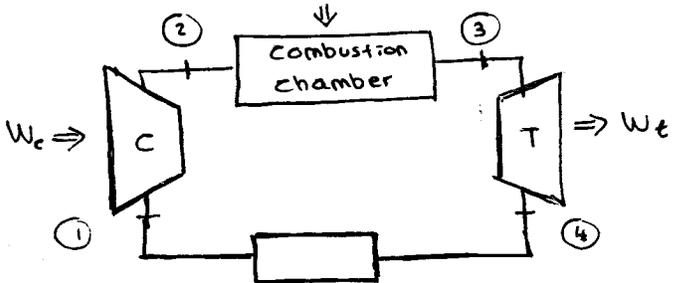
$$q_H = h_4 - h_3 = 3213.6 - 171.8 = 3041.8 \text{ kJ/kg}$$

$$\eta_{th} = \frac{W_{net,ac}}{q_H} = 29.2\%$$

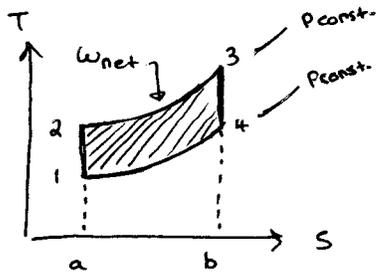
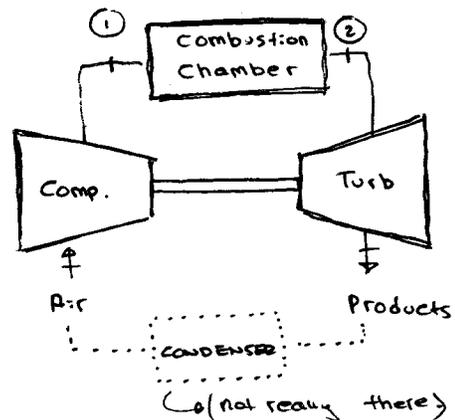
2 (From previous example)

For ideal Rankine Cycle works between same pressures, $\eta_{th} = 35.3\%$

Air Standard Power Cycles



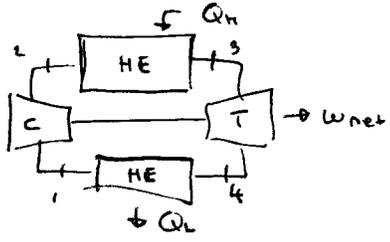
Brayton Cycle
(ideal cycle for gas)



$A_{233b} = q_H$
 $A_{414b} = q_L$

$$\eta_{th} = 1 - \frac{q_L}{q_H} = 1 - \frac{h_4 - h_1}{h_3 - h_2}$$

$$\approx 1 - \frac{C_p(T_4 - T_1)}{C_p(T_3 - T_2)} = 1 - \frac{T_1(T_4/T_1 - 1)}{T_2(T_3/T_2 - 1)}$$

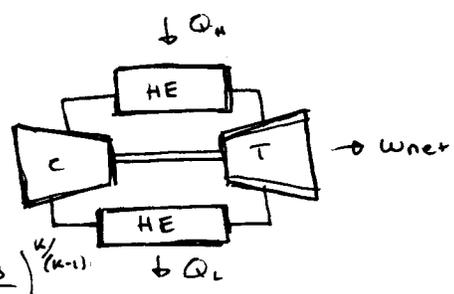


$$\frac{P_3}{P_4} = \frac{P_2}{P_1}$$

$$\left(\frac{T_2}{T_1}\right)^{\frac{\gamma}{\gamma-1}} = \left(\frac{P_3}{P_4}\right) = \left(\frac{T_3}{T_4}\right)^{\frac{\gamma}{\gamma-1}}$$

$$\frac{T_3}{T_4} = \frac{T_2}{T_1} \rightarrow \frac{T_3}{T_2} = \frac{T_4}{T_1} \rightarrow \frac{T_3}{T_2} - 1 = \frac{T_4}{T_1} - 1$$

$$\eta_{th} = 1 - T_1/T_2 = 1 - \left[\frac{1}{(P_2/P_1)^{(\gamma-1)/\gamma}} \right]$$



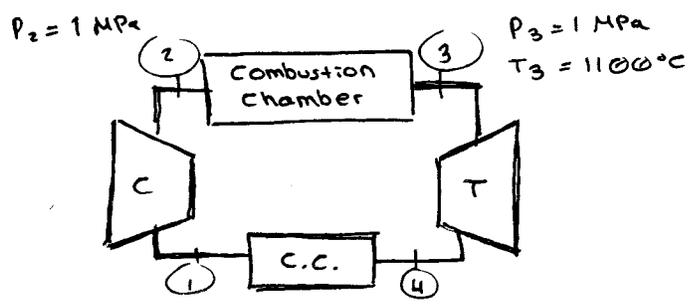
Example:

Air-standard Brayton cycle enters @ 0.1 MPa and 15°C. Leaves compressor @ 1.0 MPa.

Maximum temp is 1100°C, → Pressure + temp at each point

(in Brayton, max temp is always @

→ Comp. work
turb. work
cycle efficiency



$$\left(\frac{T_2}{T_1}\right) = \left(\frac{P_2}{P_1}\right)^{\frac{\gamma-1}{\gamma}} \Rightarrow \left(\frac{1}{0.1}\right)^{\frac{1.4-1}{1.4}}$$

$$T_2 = 288.2 \times 10^{0.286} = 556.8 \text{ K}$$

$$\left(\frac{T_4}{T_3}\right) = \left(\frac{P_4}{P_3}\right)^{\frac{\gamma-1}{\gamma}} \Rightarrow T_4 = T_3 \left(\frac{P_4}{P_3}\right)^{\gamma-1/\gamma} \rightarrow T_4 = 710.8 \text{ K}$$

$$W_c = C_p(T_2 - T_1) = (1.004 \text{ kJ/kg}\cdot\text{K})(556.8 - 288.2) = 269.5 \text{ kJ/kg}$$

$$W_t = C_p(T_3 - T_4) = (1.004)(1372.2 - 710.8) = 664.7 \text{ kJ/kg}$$

$$W_{net} = W_t - W_c = 395.2 \text{ kJ/kg} \quad ; \quad \eta_H = h_3 - h_2 = C_p(T_3 - T_2) = 819.3$$

$$\eta_{net} = (395.2) / (819.3) = 48.2 \%$$